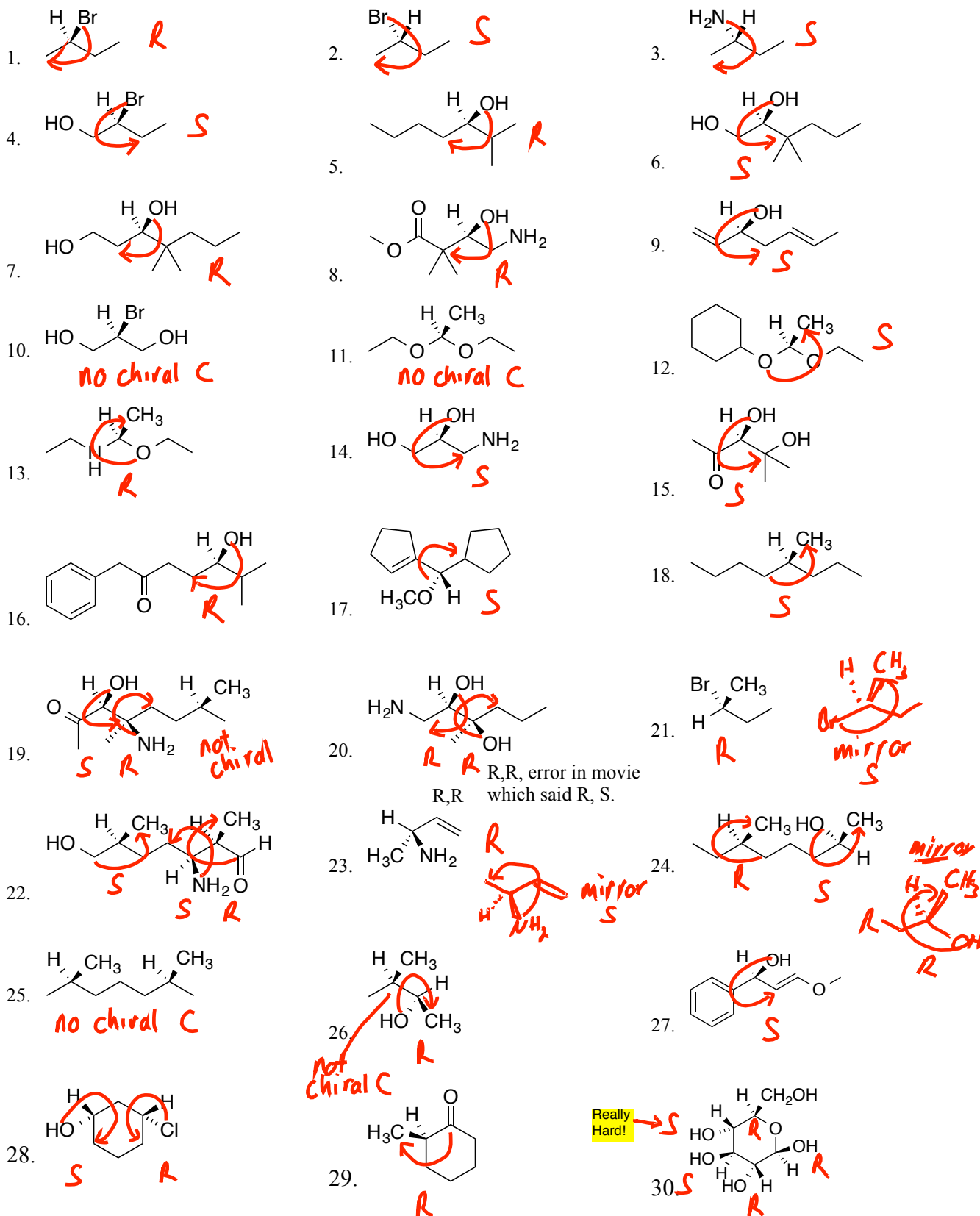


Priorities: 1. Heteroatom. 2. C-with-heteroatom.
 3. C > CH > CH₂ > CH₃.
 4. Proceed down chain until point of difference.

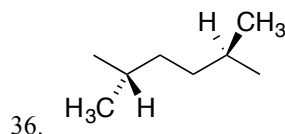
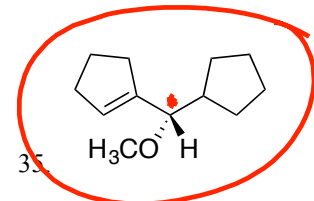
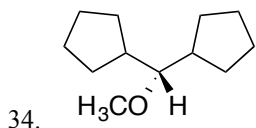
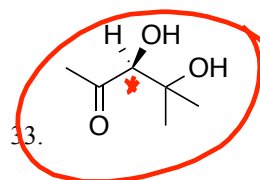
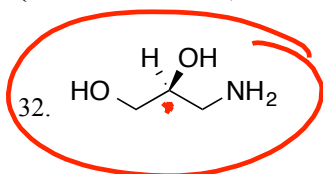
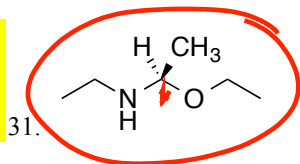
A. Designate the R/S configuration for any chiral centers in the following molecules.



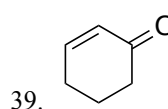
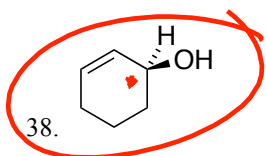
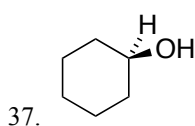
This one is very hard.
 But for each of the 5 chiral C's, the tie-breaking processes do work.
 The upper-left S is harder than the others.

**B. Identify each of the following molecule as chiral or achiral. (By circling the chiral ones.)
Write "meso" where it applies. (In other words, if it is achiral despite having chiral centers).**

One chiral carbon \Rightarrow chiral molecule

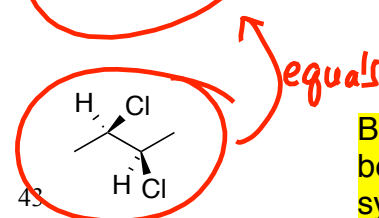
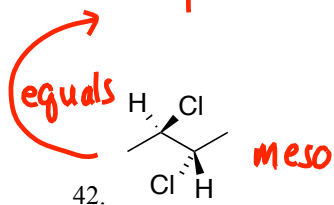
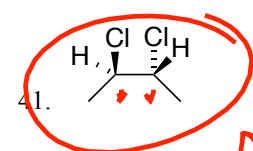
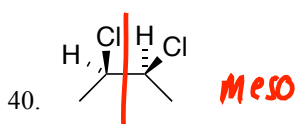


Two common attachments \Rightarrow achiral

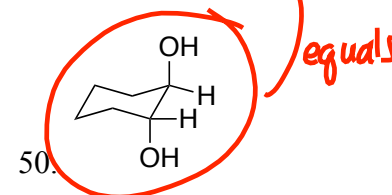
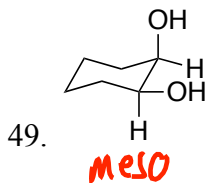
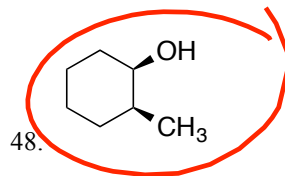
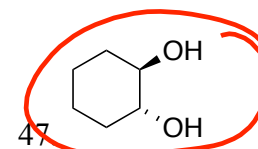
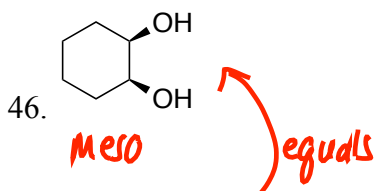
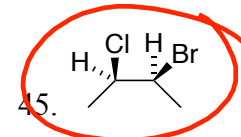
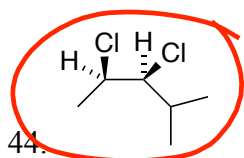


Plane of symmetry \Rightarrow achiral

Two common attachments \Rightarrow achiral



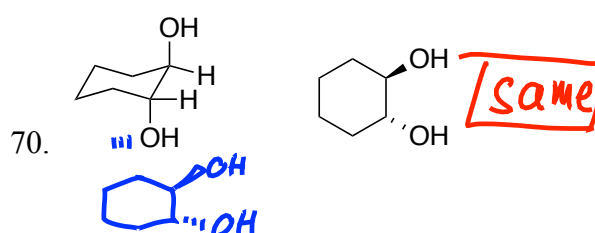
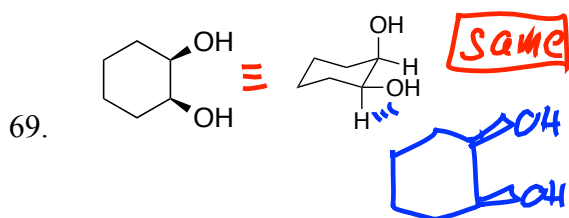
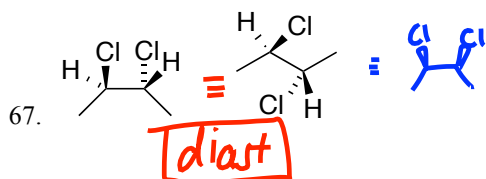
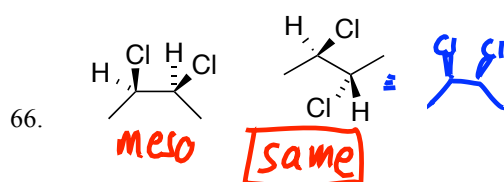
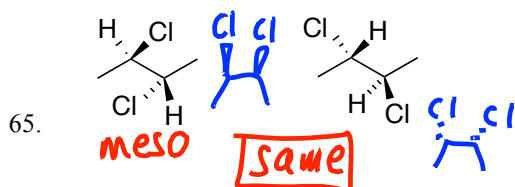
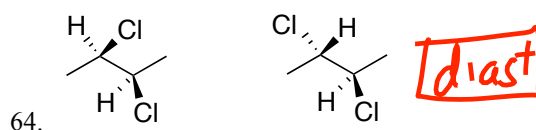
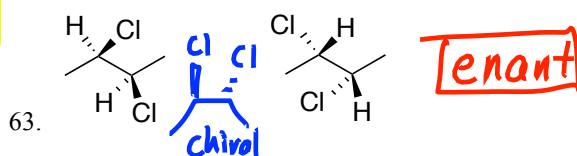
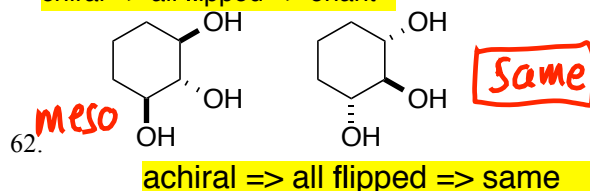
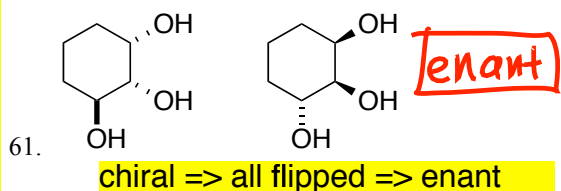
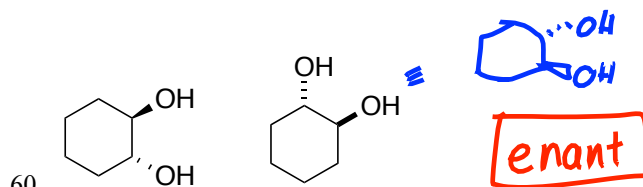
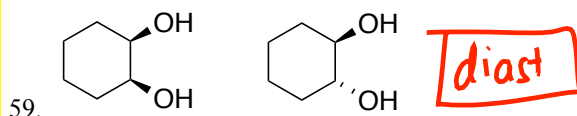
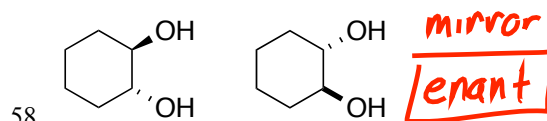
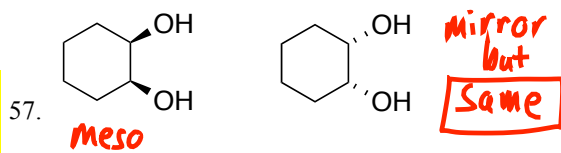
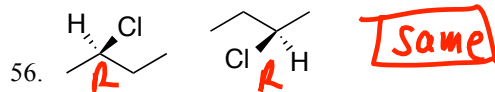
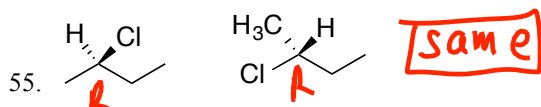
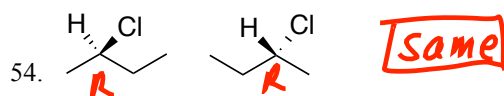
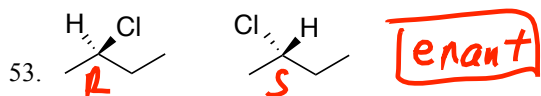
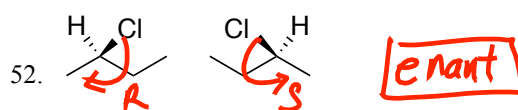
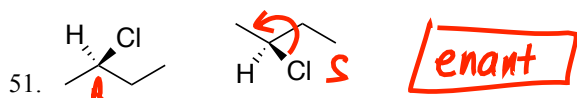
Be able to rotate in order to better visualize planes of symmetry or lack thereof



T
M
M

C. Mark the relationships between the following structures as either "same", "enantiomers", or "diastereomers".

With just one chiral C, if you can assign R/S for both, you can tell if same or different.



With two chiral C's:
1. orient them the same;
2. look for all flipped (mirror) vs some flipped (not mirror, diastereomer)
3. If mirror, check for plane of symmetry (meso, achiral, mirror same) vs not (chiral, mirror is enantiomer)